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AND CURRENT DISCOVER FILE IS DATED 23 JUNE 2008.

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L1 1753 (METAL(W) ION(W) CHROMATOGRAPHY OR METAL(W) ION(W) AFFINITY(W)  
CHROMATOGRAPHY)

=> s 11 and ammonium(w)acetate  
L2 2 L1 AND AMMONIUM(W) ACETATE

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=> dup rem 12  
PROCESSING COMPLETED FOR L2  
L2          2 DUB REM L2 (0 DUBLICATES REMOVED)
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→ dis. bib. abs. 13. 1-1

13 ANSWER 1 OF 3 CARLUS COPYRIGHT 2009 ACS on STN

L3 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2009 A  
ACCESSION NUMBER: 300E612336 CAPLUS

ACCESSION NUMBER: 2005:612336 CAPLUS  
DOCUMENT NUMBER: 143-121825

DOCUMENT NUMBER: 143:131925  
TITLE: Method for purifying EGII virus

**TITLE:** Method for purifying F3  
**INVENTOR(S):** Roger M. Hirsch

INVENTOR(S): Rossi, Mara  
PATENT ASSIGNEE(S): Rossi, Mara

PATENT ASSIGNEE(S): Ares Trading S. A., Switz.

SOURCE: PCT Int. Appl., 48 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005063811	A1	20050714	WO 2004-EP14347	20041216
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NZ, NA, NT				

NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,  
 TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW  
 RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM,  
 AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK,  
 EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT,  
 RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML,  
 MR, NE, SN, TD, TG  
 AU 2004309040 A1 20050714 AU 2004-309040 20041216  
 CA 2544333 A1 20050714 CA 2004-2544333 20041216  
 EP 1697412 A1 20060906 EP 2004-803960 20041216  
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,  
 IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK,  
 BA, HR, IS, YU  
 CN 1890265 A 20070103 CN 2004-80036591 20041216  
 BR 2004017992 A 20070427 BR 2004-17992 20041216  
 JP 2008500273 T 20080110 JP 2006-546007 20041216  
 MX 2006005584 A 20060725 MX 2006-5584 20060517  
 KR 2006135656 A 20061229 KR 2006-711610 20060613  
 US 20070129295 A1 20070607 US 2007-581172 20070206  
 PRIORITY APPLN. INFO.: EP 2003-104925 A 20031222  
 WO 2004-EP14347 W 20041216

**AB** The invention provides a method for purifying recombinant human FSH or an FSH variant, comprising the steps: (1) ion exchange chromatog.; (2) immobilized metal ion chromatog.; (3) hydrophobic interaction chromatog. which may be carried out in any order.  
**REFERENCE COUNT:** 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN  
 ACCESSION NUMBER: 1991:451273 CAPLUS  
 DOCUMENT NUMBER: 115:51273  
 ORIGINAL REFERENCE NO.: 115:8913a,8916a  
**TITLE:** Investigation of a cation exchange separation method for the determination of transition metal ions in anaerobic sealants  
**AUTHOR(S):** O'Dea, Philip; Deacon, Marian; Smyth, Malcolm R.; Leonard, Raymond G.  
**CORPORATE SOURCE:** Sch. Chem. Sci., Dublin City Univ., Dublin, Ire.  
**SOURCE:** Analytical Proceedings (1991), 28(3), 82-4  
**CODEN:** ANPRDI; ISSN: 0144-557X  
**DOCUMENT TYPE:** Journal  
**LANGUAGE:** English  
**AB** The applicability of ion-exchange chromatog. to anaerobic sealants poly[(ethylene glycol) di(2-ethylhexanoate)] for the determination of transition metal ion content was demonstrated with detection by post-column derivatization using the color-forming monosodium salt of 4-(2-pyridylazo)resorcinol. The sample was dissolved in chloroform, extracted with aqueous HCl, the acid extract passed through a C18 separatory column prior to being injected onto the ion-exchange separation system, and was then subjected to the above-cited derivatization. Linear calibration curves were obtained for Zn<sup>2+</sup>, Co<sup>2+</sup>, Fe<sup>2+</sup>, and Mn<sup>2+</sup> over 1.0-10 ppm and for Cu<sup>2+</sup> from 2-9 ppm. The limit of detection based on the peak heights, being greater than 3 times the base line noise, was approx. 0.5 ppm for Zn<sup>2+</sup>, Co<sup>2+</sup>, Fe<sup>2+</sup>, and Mn<sup>2+</sup>, and 1 ppm for Cu<sup>2+</sup>.

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 COST IN U.S. DOLLARS  
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SINCE FILE ENTRY	TOTAL SESSION
30.37	30.59

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CA SUBSCRIBER PRICE	ENTRY	SESSION
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